

# Double Chiral Induction Enables a Stereoselective Carbonyl Allylation with Simple Alkenes under the Sequential Catalysis of Palladium Complex and Chiral Phosphoric Acid

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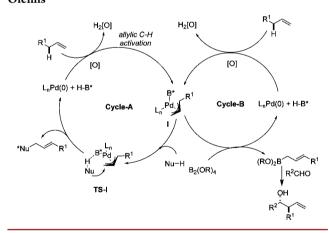
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Supporting Information

ABSTRACT: An enantioselective carbonyl allylation of aldehydes with simple alkenes has been achieved via a one-pot protocol consisting of a Pd-catalyzed allylic C-H borylation with bis(pinanediolato)diboron and a chiral Brønsted acid catalyzed asymmetric allylborylation, delivering homoallylic alcohols in high yields and with excellent diastereo- and enantioselectivities. The double chiral induction of chiral allylic borate and chiral phosphoric acid allows the reaction to give excellent stereoselectivities.

rminal alkenes are an easily accessible and abundant feedstock, and are also among the most useful starting materials in organic synthesis, attributable to the rich chemistry capable of occurring at the carbon-carbon double bond. In recent decades, the direct functionalization of allylic C-H bond of terminal alkenes has received a great deal of interest, 2 leading to the frequent appearance of new protocols.<sup>3,4</sup> In contrast, only a few successful examples describe catalytic, enantioselective allylic C-H functionalization reactions with the exception of the asymmetric allylic C-H oxidation catalyzed by chiral copper complexes. White and co-workers found that the combination of a palladium complex and a chiral Lewis acid allowed a direct, asymmetric esterification of terminal olefins, but with modest enantioselectivity. Trost demonstrated that the chiral phosphoramidite ligands were able to offer good enantioselectivity in the Pd-catalyzed asymmetric allylic alkylation of 1, 3-diketones with terminal alkenes. Recently, our group established a variety of enantioselective allylic C-H functionalization reactions by virtue of cooperative catalysis of palladium and Brønsted acids. These reactions basically proceed via a cascade allylic C-H oxidation and enantioselective substitution of the  $\eta^3$   $\pi$ allylpalladium species I with different nucleophiles (Cycle-A, Scheme 1). On the other hand, the  $\pi$ -allylpalladium species I is principally able to react with diboron compounds to generate allylic boron intermediates, which are able to undergo allylation reactions with electrophiles (Cycle-B, Scheme 1). However, such a principally possible transformation was not realized until recently when Szabó and co-workers 10 found that exocyclic alkenes were able to undergo a direct carbonyl allylation reaction of aldehydes, presumably proceeding via an allylic C-H borylation (eq 1, Scheme 2). Very recently, we disclosed a

# Scheme 1. Pd-Catalyzed C-H Functionalization of Terminal **Olefins**



diastereoselective palladium-catalyzed carbonyl allylation of aldehydes directly using simple acyclic olefins as allylating reagents, which was preliminarily thought to proceed via the reaction pathway shown in Cycle-B in Scheme 1.11 However, previous trials to establish an asymmetric version only led to moderate enantioselectivity (eq 2, Scheme 2). Herein, we describe a one-pot Pd/chiral phosphoric acid sequentially catalyzed process <sup>12</sup> consisting of an allylic C–H borylation and enantioselective allylation of aldehydes by using a double chiral

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# Scheme 2. Previous Reports on Allylation of Aldehydes with Olefins

induction strategy, to allow the formation of *anti*-homoallylic alcohols in high enantioselectivities.

Because the **CPA1** that has been assumed to give high levels of stereochemical outcome in asymmetric allylborylation of aldehydes<sup>13</sup> only gave a moderate enantioselectivity and yield, some other sterically demanding chiral phosphoric acids were again evaluated for the reaction under otherwise identical conditions (Figure 1). Unfortunately, none of the phosphoric

**Figure 1.** Evaluation of chiral phosphoric acids in the one-operation procedure.

acids was able to deliver satisfactory enantioselectivity, while CPA1 and CPA4 showed superior enantioselectivity. Basically, the asymmetric allylborylation reactions of aldehydes proceeding at a lower temperature gave a higher stereochemical outcome, but in contrast, the Pd-catalyzed allylic C-H borylation requires a relatively higher reaction temperature. To balance the reaction conditions of two individual steps, a one-pot sequential process was investigated. Indeed, an enhanced enantioselectivity of 75% ee was obtained from the sequential reaction going through the Pd and phosphoric acid CPA1 cooperatively catalyzed allylic C-H borylation and Brønsted acid catalyzed allylborylation of 4nitrobenzaldehyde (Table 1, entry 1). Elevating the allylic C-H borylation reaction temperature to 50 °C while maintaining the allylation reaction at room temperature led to a higher yield, but a considerably lower enantioselectivity was observed (entry 2). Interestingly, the use of CPA4 to replace CPA1 resulted in a moderate yield, but with 81% ee under otherwise identical reaction conditions (entry 3). Double chiral induction has been considered to be a powerful strategy for the development of stereoselective reactions. 14 Thus, bis(pinanediolato)diboron 2b, 10b,15 a commercially available chiral reagent, was tested and found to give product 4a in 49% yield and with 91% ee (entry 4). In contrast, the use of the enantiomer of CPA4 as a cocatalyst

Table 1. Evaluation of Catalyst Systems and Optimization of Reaction Conditions<sup>a</sup>

entry	CPA	2	$t_1$ (°C)	<i>t</i> <sub>2</sub> (°C)	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	CPA1	2a	25	25	62	75 <sup>d</sup>
2	CPA1	2a	50	25	81	60 <sup>d</sup>
3	CPA4	2a	50	25	49	81
4	CPA4	2b	50	25	44	91
5	ent-CPA4	2b	50	25	48	53 <sup>d</sup>
6		2b	50	25	75	18
7	CPA4	2b	50	-10	34	92
8	CPA4	2b	50	-20	33	93
9 <sup>e</sup>	CPA4	2b	50	-20	72	93
$10^{e,f}$	CPA4	2b	50	-20	85	93

 $^a$ Unless indicated otherwise, the reaction of 1a (0.2 mmol), 2 (0.2 mmol), Pd(dba) $_2$  (0.01 mmol), PPh $_3$  (0.02 mmol), CPA (0.02 mmol), and N-fluorobenzenesulfonimide (NFSI) (0.2 mmol) was carried out in toluene (1.5 mL) at 50 °C for 24 h, and then 4-nitrobenzaldehyde 3a (0.1 mmol) was added; resultant mixture was stirred for an additional 24 h.  $^b$ Isolated yield with >20:1 dr.  $^c$ Determined by chiral HPLC.  $^d$ An opposite enantiomer of 4aa was obtained.  $^c$ 0.3 mmol of 2b was employed.  $^f$ CPA was added to the reaction of 3a after the allylic borylation reaction.

allowed the reaction to give an opposite enantiomer with a much eroded enantioselectivity (entry 5). In addition, the use of bis(pinanediolato)diboron 2b as a sole chiral element provided a much diminished stereochemical outcome (entry 6). These results indicated that the double chiral induction indeed existed and the (R)-configuration of chiral phosphoric acid matches the chirality of 2b to induce higher enantioselectivity. Although with a reduced yield, performing the allylation step at -20 °C did provide a slightly enhanced enantioselectivity of 93% ee (entry 8). When the ratio of chiral diboronate **2b** to **1a** was tuned to 3:2, the sequential reaction proceeded more cleanly to deliver a 72% yield, without sacrificing the enantioselectivity (entry 9). Interestingly, the desired product 4aa was isolated in a much improved yield and with a maintained enantioselectivity by introducing the Brønsted acid catalyst CPA4 after the borylation step completely consumed the allylbenzene 1a (entry 10).

With the optimal conditions in hand, the substrate scope of allylbenzene derivatives was examined. A variety of allylbenzene derivatives underwent the Pd/Brønsted acid sequentially catalyzed allylic borylation and asymmetric carbonyl allylation reaction to give corresponding products 4 in good yields and with excellent diastereo- and enantioselectivities (Table 2). In general, the substitution pattern of the benzene ring exhibited a subtle effect on either the yield or stereochemical outcome, as illustrated by reactions to generate chiral homoallylic alcohols 4ba to 4da. Both electron-donating and -withdrawing substituents on the allylbenzene provided the desired products with satisfactory yields and stereoselectivity (4ea-4ga). In addition, 1-allyl- and 2-allylnaphthalenes were both well tolerated, delivering the desired products in good yields and with good enantioselectivities (4ha and 4ia). The substrates with aryl substituents on the benzene ring could also be transformed into

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#### Table 2. Substrate Scope of Allylbenzene<sup>a</sup>

#### 4 (yieldb, drc, eed)

<sup>a</sup>Unless indicated otherwise, the reaction of 1 (0.2 mmol), 2b (0.3 mol),  $Pd(dba)_2$  (0.01 mmol),  $PPh_3$  (0.02 mmol), and NFSI (0.2 mmol) was carried out in toluene (1.5 mL) at 50 °C for 24 h, and then 4-nitrobenzaldehyde 3a (0.1 mmol) and CPA4 (0.02 mmol) were added; the resultant mixture was stirred at -20 °C for an additional 24 h. <sup>b</sup>Isolated yield. <sup>c</sup>The ratio of anti/syn was determined by <sup>1</sup>H NMR of the crude products. <sup>d</sup>Determined by chiral HPLC.

the products in moderate yields and good enantioselectivities (**4ja** and **4ka**). However,  $\alpha$ -methylstyrene underwent a very slow reaction to give product **4la** in only 10% yield and with 23% ee. The absolute configuration of **4fa** was assigned to be (1*R*,2*S*) by X-ray analysis of its single crystal (see Supporting Information).

In addition, the protocol was amenable to a wide variety of aromatic aldehydes delivering the desired products in good to high yields and with excellent diastereo- and enantioselectivities (Table 3). However, the substitution pattern of benzaldehydes has considerable impact on the stereochemical control. Basically, both 3- and 4-substituted benzaldehyde derivatives were generally well tolerated (4ac-4ah). In contrast, 2-nitrobenzaldehyde underwent a much slower reaction and gave a diminished enantioselectivity in comparison with either 3- or 4nitrobenzaldehydes (4ab vs 4ac and 4aa in Table 2). Neither the electronic property nor substitution pattern exhibited an impact on the diastereoselectivity, while electron-deficient benzaldehydes generally gave higher yields and enantioselectivities than the electron-rich ones (4ac, 4ad and 4af vs 4ae and 4ah). In addition to the benzaldehydes, other aryl aldehydes were also compatible for the transformation (4ai-4al). An aliphatic

Table 3. Substrate Scope of Aldehydes<sup>a</sup>

4 (yieldb, drc, eed)

 $^a\mathrm{Unless}$  indicated otherwise, the reaction of 1a (0.2 mmol), 2b (0.3 mol), Pd(dba) $_2$  (0.01 mmol), PPh $_3$  (0.02 mmol), and NFSI (0.2 mmol) was carried out in toluene (1.5 mL) at 50 °C for 24 h, and then the aldehydes 3 (0.1 mmol) and CPA4 (0.02 mmol) were added; the resultant mixture was stirred at -20 °C for an additional 24 h.  $^b\mathrm{Isolated}$  yield.  $^c\mathrm{The}$  ratio of anti/syn was determined by  $^1\mathrm{H}$  NMR of the crude products.  $^d\mathrm{Determined}$  by chiral HPLC.

aldehyde, cyclohexanecarboxaldehyde, also turned out to be a suitable substrate for the reaction, providing **4am** in an 83% yield and with 89% enantiomeric excess.

In conclusion, we have established a one-pot protocol consisting of a Pd-catalyzed allylic borylation with a chiral diborate and a chiral Brønsted acid catalyzed asymmetric allylation of aldehydes, delivering homoallylic alcohols in high yields and with excellent diastereo- and enantioselectivities, providing an efficient method for the stereoselective functionalization of allylic C—H bonds. It turned out that the double chiral induction of chiral allylic borate and chiral phosphoric acid allows the reaction to exhibit high enantioselectivity.

### ASSOCIATED CONTENT

# Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03378.

Experimental details and characterization data (PDF)

X-ray data for compound 4fa (CIF)

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#### **Notes**

The authors declare no competing financial interest.

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